

## SI Units, Accuracy and Precision

### SI (Systeme International) Base units

- Length = m (meters)
- Mass = kg (any version of a gram is ok)
- Time = s (seconds)
- Electric current = A (amperes)
- Temperature = K (Kelvin)
- Amount of a substance = mole (mol)
- And just for fun –  
Luminous intensity = candela (cd)

## SI Units Continued

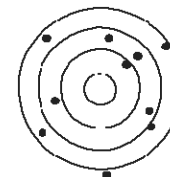
### SI Prefixes

- Tera (T) = trillion ( $10^{12}$ )
- Giga (G) = billion ( $10^9$ )
- Mega (M) = million ( $10^6$ )
- Kilo (k) = thousand ( $10^3$ )
- Hecto (h) = hundred ( $10^2$ )
- Deca (da) = ten ( $10^1$ )
- -----BASE UNIT-----
- Deci (d) = tenth ( $10^{-1}$ )
- Centi (c) = hundredth ( $10^{-2}$ )
- Milli (m) = thousandth ( $10^{-3}$ )
- Micro ( $\mu$ ) = millionth ( $10^{-6}$ )
- Nano (n) = billionth ( $10^{-9}$ )
- Pico (p) = trillionth ( $10^{-12}$ )

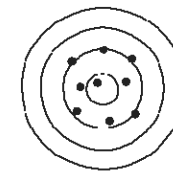
## Accuracy and Precision

- Which is better – to be accurate or precise? (ponder for a moment)

## Accuracy and Precision

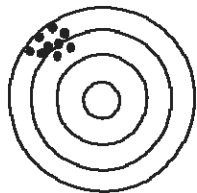


Low Accuracy,  
Low Precision

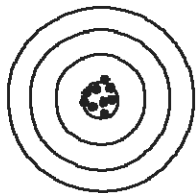


High Accuracy,  
Low Precision

## Accuracy and Precision



Low Accuracy,  
High Precision



High Accuracy,  
High Precision

## Accuracy and Precision

- Precision is the reproducibility of the measurements. Precise measurements all end up very close in value.
- Accuracy is the closeness of a measurement to the true value.
- Which is better? To be accurate or precise?

## Sig Digs!!! How many?

1. All non-zero numbers are always significant.  
Example: 345 has 3 significant digits
2. Captive zeros (zeros between non-zero numbers) are always significant.  
Example: 2.004 has 4 significant digits
3. Trailing zeros (zeros to the right of a non-zero number) are significant only when the number contains a decimal point.  
Example: 300.0 has 4 significant digits, but 300 has only 1 significant digit
4. Leading zeros (zeros to the left of the first non-zero number) are not significant.  
Example: 0.00429 has 3 significant digits
5. Exact numbers are considered to have unlimited significant digits. Numbers such as those determined by counting, or from definitions are considered exact.  
Examples: 3 pennies – the 3 has unlimited significant digits  
1 foot = 12 inches – the 1 and 12 have unlimited significant digits, based on the definition of a foot

## Calculating with Sig Digs

**Multiplication and Division:** Your answer can only have as many significant digits as the factor with the *least* number of significant digits.  
Example:  $29.1 \text{ cm} \times 1.04 \text{ cm} \times 22 \text{ cm} = 665.808 \text{ cm}^3$  (on calculator), but must be rounded to 2 significant digits which equals  $670 \text{ cm}^3$ .

**Addition and Subtraction:** Your answer must be rounded to the decimal place of the number that had the *least* number of decimal places.  
Example:  $145.92 \text{ g} + 22.0 \text{ g} + 57.712 \text{ g} = 225.632 \text{ g}$  (on calculator), but must be rounded to the tenths place because 22.0 is only to the tenths place. Thus, the answer is changed to 225.6 g.

Remember – multiplication and division go by total number of digits; addition and subtraction go by least decimal place.